

Thermodynamics and Ion Transport of Multivalent Cathode Materials

Pieremanuele Canepa^{a,b}

^aDepartment of Materials Science and Engineering, National University of Singapore, 117575 Singapore

^bMaterials Science Division, Lawrence Berkeley National Laboratory, Berkeley, CA 94720, USA

Abstract: Multi-valent (MV) batteries that replace Li^+ ions with inexpensive MV cations, including Mg^{2+} , Zn^{2+} and Ca^{2+} represent a promising approach to meet the high energy density requirements of the next generation of electrical devices.^[1] Perhaps the most pressing challenge in achieving high energy density MV-ion systems is to develop suitable cathode materials and conductors with a significant voltage and high MV-ion transport.^[1-2] To date, there have been limited examples demonstrating the feasibility of rechargeable MV batteries, and among them, most of the focus has been on Mg technology. From the limited experimental studies performed so far, the feasibility of a battery technology based on MV intercalation is not yet clear. The cathode represents a critical component of this technology. Therefore, it is crucial to assess the feasibility of MV cathodes. I will present a detailed analysis, based on first-principles DFT calculations, of MV intercalation in promising candidates, including the spinel MVB_2X_4 system (with B = transition metal and X the anion) and the polymorphs of the layered vanadium pentoxide (V_2O_5).^[1-7] I will demonstrate that computational materials science is a powerful tool to pave the successful development and optimization of new materials for energy dense MV batteries.

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