

Structural investigations of orthorhombic V_2O_5 nanowires in the full cell with magnesiated Mo_6S_8 anode

Qiang Fu^{a,*}, Angelina Sarapulova^a, Vanessa Trouillet^{a,c}, Lihua Zhu^a, Francois Fauth^d, Stefan Mangold^e, Edmund Welter^f, Sylvio Indris^{a,b}, Michael Knapp^{a,b}, Sonia Dsoke^{a,b}, Natalia Bramnik^a, Helmut Ehrenberg^{a,b}

^aInstitute for Applied Materials (IAM), Karlsruhe Institute of Technology (KIT), Karlsruhe, Germany.

^bHelmholtz Institute Ulm for Electrochemical Energy Storage (HIU), Ulm, Germany. ^cKarlsruhe Nano Micro Facility (KNMF), Karlsruhe Institute of Technology (KIT), Germany. ^dCELLS-ALBA Synchrotron, Barcelona, Spain. ^eInstitute for Photon Science and Synchrotron Radiation, Karlsruhe Institute of Technology (KIT), Karlsruhe, Germany. ^fDeutsches Elektronen-Synchrotron DESY, Hamburg, Germany

Email: qiang.fu@kit.edu (Q. Fu)

The V_2O_5 cathode is known to be good material for lithium and magnesium insertion[1,2]. Therefore, it is a possible candidate as cathode material in magnesium-ion batteries. As known, the lithium insertion into V_2O_5 has already been very well studied[1]. However, only limited work focuses on the structural investigation upon Mg insertion into V_2O_5 . For example, Gershinsky et al. [3] reported that *ex situ* XRD confirmed the insertion of Mg into the crystal structure with a reversible capacity of $\sim 150 \text{ mAh g}^{-1}$ for V_2O_5 thin film in the potential range 2.2 -3.0 V vs. Mg^{2+}/Mg . The detailed structural investigation of bulk V_2O_5 cathode in the whole insertion and extraction processes is still missing.

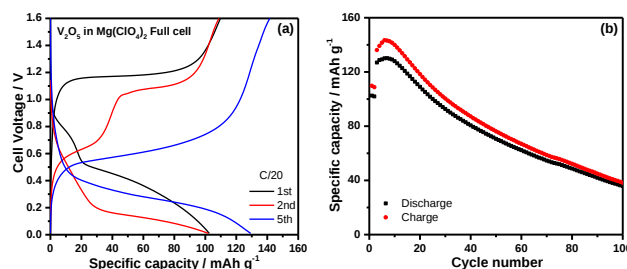


Figure 1: Discharge-charge profiles (a) and cycling performance (b) of $V_2O_5 \mid Mg(ClO_4)_2/AN \mid Mg_{x/2}Mo_6S_8$ ($x \sim 2$) full cell (C/20) in the voltage range of 1.6-0.01 V.

In the present work, orthorhombic V_2O_5 Nanowires were successfully synthesized via a hydrothermal method. A cell was first made to obtain the pre-magnesiated Mo_6S_8 anode due to the incompatibility of metallic Mg with chosen "standard" electrolyte. A full cell system was built using V_2O_5 as cathode, 1M $Mg(ClO_4)_2$ in acetonitrile (AN) and $Mg_{x/2}Mo_6S_8$ ($x \sim 2$) anode. The V_2O_5 nanowires deliver an initial discharge/charge capacity of $100 \text{ mAh g}^{-1}/114 \text{ mAh g}^{-1}$ and the highest discharge capacity of 128 mAh g^{-1} in the 5th cycle at C/20 rate in the full cell [4].

The structural evolution and oxidation state and local structural changes and the reversibility of the Magnesium insertion/extraction in the V_2O_5 were studied in the above mentioned full-cell system. The formation of the new phase Mg-rich $Mg_xV_2O_5$ during Mg insertion and the recovery of V_2O_5 during Mg extraction were identified via *in situ* synchrotron diffraction and *ex situ* Raman. The reduction/oxidation of vanadium during the Mg insertion/extraction was confirmed by *in situ* X-ray absorption spectroscopy (XAS) and *ex situ* XPS.

References

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